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Chemical Names And Formulas Chapter

Chapter 9 - Chemical
Names and Formulas.

STUDY. Flashcards.

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Test. PLAY. Match.

Gravity. Created by.

bibiloni1901. Terms in

this set (20) Monatomic

Ion. Consists of a single

atom with a positive or

negative charge

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resulting from the loss
of gain of one or more
valence electrons and
behaves as a unit, yet
it still has a ...

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Writing Ionic

Compound Formulas

Example: Ammonium
sulfate (note the 2

word name) 1. Write
the formulas for the
cation and anion,
including CHARGES!

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Check to see if charges are balanced. 3.

Balance charges, if necessary, using subscripts. Use parentheses if you need more than one of a polyatomic ion. Use the criss-cross

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Naming Ions &
Counting Atoms:
Chapter 9.1 Naming &
Writing Formulas for
Ionic Compounds:
Chapter 9.2 Naming &
Writing Formulas for
Molecular Compounds:
Chapter 9.3 Naming &

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Writing Formulas for
Acids: Chapter 9.4
Quizlet Chapter 9
vocabulary Answer
Keys Ionic Compound
Naming and Formula
Examples from notes

Chapter 9: Chemical Names & Formulas

chapter 9 chemical
names and formulas
Flashcards and Study
Sets | Quizlet. 1. write
the element with the
lower electronegativity
first (g.... 1. write the

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element with the lower electronegativity first (g.... 1. the oxidation number of an ionic compound is equal to its c.... 1. write the element with the ...

chapter 9 chemical names and formulas Flashcards and Study ...

Chapter 9 - Chemical Names and Formulas.
Jennie L. Borders.
Periodic Table for Naming. Section 9.1

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Regular Metals Review.

The regular metals include Group 1 (except H) and 2 and aluminum. When naming a compound that starts with a regular metal, name the first element and add -ide to the second element (except for polyatomic ions).

Chapter 9 - Chemical Names and Formulas

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Answers Quiz

NAME: CLASS: BLM

2-14 CHAPTER 4

Chemical Names and
Formulas Goal Review
your understanding of
chemical names and
formulas. What to Do
Complete the following
table. Chemical
Formula Ionic or
Covalent? 3 Name of
Compound Iovtcl.n.
Qvn mo potassium
sulphide calcium oxide
ammonium chloride

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mat zinc oxide oxygen
dichloride CO_2 carbon
tetrabromide

KM 754e-201704261 52951

Chapter 9 Chemical
Names and Formulas 85
Names and Formulas
for Bases (page 273) 6.

A base is a compound
that produces _____
when dissolved in
water. 7. How are

bases named?

SECTION 9.5 THE LAWS
GOVERNING

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**FORMULAS AND
NAMES** (pages
274-279) This section
uses data to
demonstrate that a
compound obeys the
law of definite
proportions.

Name Date Class CHEMICAL NAMES AND FORMULAS 9

Chapter 6 Chemical
Names and Formulas.
Compounds composed
of two elements.
Compounds composed

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of metal cations and....
Atoms or groups of
atoms with a negativ....
Atoms or groups of
atoms that that have....
Binary Compounds.
Ionic Compounds.
Anions. Ions or
Polyatomic Ions.

formulas chapter 6 chemical names Flashcards and Study ...

Chemical Names and
Formulas 281 CHAPTER
9 Assessment 42. a. 2-

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b. 1+ c. 1- d. 3+ 43. a.

2+ b. 2+ c. 3+ d. 1+

44. a. barium ion b.
iodide ion c. silver ion
d. mercury(II) ion 45.

cyanide, CN^- and
hydroxide, OH^- 46. a.

hydroxide ion b.
lead(IV) ion c. sulfate
ion d. oxide ion 47.

zero; A compound is
electrically neutral. 48.

The symbols for the
cation and

CHAPTER 9 Study Guide - Quiz

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2. Write the formulas for the compounds formed from these pairs of ions. c. Ca^{2+} K^+ e. O^{2-} Cu^{+} h. Sn^{2+} Cl^-

3. Name the following binary ionic compounds. a. MnO_2 b. CaCl_2 d. SrBr_2 e. f.

Chapter 9 Chemical Names and Formulas
221

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...

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Chapter 9 Chemical Names and Formulas

223 2 Complete this table by writing correct formulas for the compounds formed by combining positive and negative ions 3 Name the following compounds a K_3PO_4 c $NaHSO_4$ e N_2O_5 g PI_3 b $Al(OH)_3$ d HgO f NBr

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Chemical Names And
Formulas Chapter 9

Quiz Chemical Names
And Formulas Chapter

Unit 5: Chemical

Names & Formulas

Chemical Names &

Formulas (Chapter 9) 2

3 Monatomic Ion

Formation &

Nomenclature

Monatomic Ions consist

of a _____ atom with a

positive or negative

charge resulting from

the gain or loss of one

or more valence _____

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Cation: any atom or

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Formulas Chapter 9
Quiz**

Chemical Names and
Formulas 281 CHAPTER
9 Assessment 42 a 2- b
1+ c 1- d 3+ 43 a 2+ b
2+ c 3+ d 1+ 44 a
barium ion b iodide ion
c silver ion d
mercury(II) ion 45
cyanide, CN^- and
hydroxide, OH^- 46 a
hydroxide ion b

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lead(IV) ion c sulfate
ion d oxide ion 47 zero;
A compound is
electrically neutral 48
The symbols for the
cation and

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...

Chapter 9 Chemical
Names and Formulas
221 Name Date 4 Write
formulas for the
following ionic
compounds a sodium

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phosphate c sodium

hydroxide e

ammonium chloride b

magnesium d

potassium cyanide 5

Write formulas for
compounds formed
from these pairs of ions
b , 6 Name the
following compounds ...

[MOBI] Chapter 9 Chemical Names And Formulas Answers

Chemical Names and
Formulas 1. Chapter 9
Naming and Writing

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Formulas 2. Classifying Compounds The

The system for naming an ionic compound is different from that for naming a covalent compound, so before a compound can be named, it must be classified as ionic or covalent.

Chemical Names and Formulas - SlideShare

Section 1 Chemical
Names and Formulas

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And Formulas Chapter 7 Naming Binary Ionic Quiz

Compounds, continued
Compounds Containing
Polyatomic Ions • Many
common polyatomic
ions are oxyanions
—polyatomic ions that
contain oxygen. •
Some elements can
combine with oxygen
to form more than one
type of oxyanion. •
example: nitrogen can
form or.

Section 1 Chemical

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Names and Formulas Chapter 7 Naming ...

Chapter 7: Chemical
Formulas and Chemical
Compounds Section
7-1: Chemical Names
and Formulas 7-1-1

Explain the significance
of a chemical formula
The subscript indicates
that there are 8 carbon
atoms in the molecule
of octane The subscript

[EPUB] Chapter 7 Chemical Formulas And Compounds

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Cacodyl, a compound
containing arsenic, was
reported in 1842 by
the German chemist
Robert Wilhelm
Bunsen.

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